

# Nitrate analysis of water using the quartz fiber optics dip probe on the Cary 50/60 UV-Vis

## Application Note

### Environmental

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#### Introduction

Environmental laboratories analyze thousands of water samples a year to determine the concentration levels of heavy metals and other ions, such as nitrates, phosphates and fluorides. To increase sample throughput and efficiency, optical fibers may be used to measure the absorbance of the sample. This allows for analysis on- or off-site, which is more appealing than a conventional cuvette. This paper presents and discusses results obtained from measuring the nitrate content in water using the quartz dip probe on the Cary 50 UV-Vis spectrophotometer. This experiment can also be done on the Cary 60 UV-Vis.

#### Experimental

##### Equipment

- Cary 50 (Cary 60)UV-Vis spectrophotometer
- Dip probe fiber optics coupler
- Quartz fiber optic dip probe
- Cary WinUV software

##### Reagents

- Potassium nitrate (A.R.)
- 37% m/v Hydrochloric acid (A.R.)
- Chloroform (A.R.)
- Water - distilled and de-ionized



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## Method

The experimental procedure was taken from *Standard Methods for the Examination of Water and Wastewater*<sup>1</sup> and is also described in *UV Instruments At Work No. 59*<sup>2</sup>. In brief, standard solutions were prepared in the concentration range of 0 - 7mg NO<sub>3</sub><sup>-</sup> - N/L and the absorbance measured at 220 and 275 nm. The measurement at two wavelengths allows correction for the interference due to dissolved organic matter, by calculating the difference between both absorbance readings (Equation 1).

$$\text{Abs}(220) - 2 \times \text{Abs}(275) \quad \text{Equation 1}$$

The application used was the Cary WinUV Concentration software which evaluates the result of  $\text{Abs}(220) - 2 \times \text{Abs}(275)$  dynamically as a function of concentration. The following instrument settings were used for data collection.

## Instrument Settings

|                               |                         |
|-------------------------------|-------------------------|
| User Result                   | = Read(220)-2*Read(275) |
| Ordinate Mode                 | Abs                     |
| Ave Time (sec)                | 1.0000                  |
| Replicates                    | 3                       |
| Standard/Sample averaging     | OFF                     |
| Weight and volume corrections | OFF                     |
| Fit type                      | Quadratic               |
| Min R <sup>2</sup>            | 0.95000                 |
| Concentration units           | mg/L                    |

## Results

Figure 1 shows the calibration curve obtained using the Quartz Fiber Optics Dip Probe. The Y axis, Abs, is the resultant from Equation 1 and the X-axis is the concentration of Nitrate Standards in mg/L.

A quadratic function, Equation 2, was fitted to 6 standards giving a correlation coefficient of 0.99931. The raw absorbance data and statistics for the calibration standards are shown in Table 1.

$$\text{Abs} = -0.00017\text{conc}^2 + 0.23364\text{conc} + 0.01705 \quad \text{Eq. 2}$$

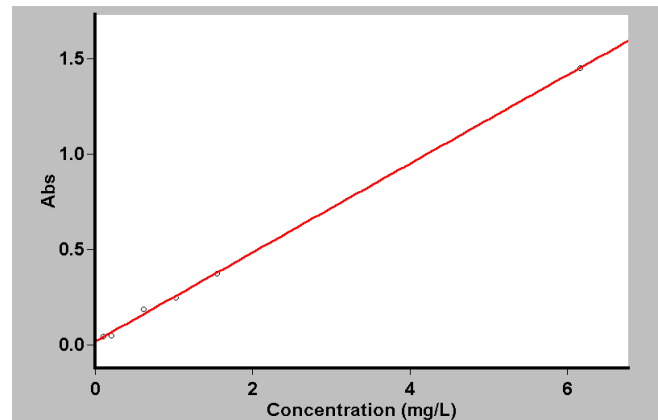


Figure 1. Calibration curve with quadratic fit

Table 1. Nitrate standards data for calibration curve

| Std   | Conc mg/L | Mean Abs | SD     | %RSD | Raw Abs |
|-------|-----------|----------|--------|------|---------|
| Std 1 | 0.103     | 0.0443   | 0.0036 | 8.03 | 0.0472  |
|       |           |          |        |      | 0.0404  |
|       |           |          |        |      | 0.0454  |
| Std 2 | 0.205     | 0.0488   | 0.0002 | 0.37 | 0.0488  |
|       |           |          |        |      | 0.0490  |
|       |           |          |        |      | 0.0487  |
| Std 3 | 0.616     | 0.1856   | 0.0011 | 0.57 | 0.1856  |
|       |           |          |        |      | 0.1846  |
|       |           |          |        |      | 0.1867  |
| Std 4 | 1.027     | 0.2467   | 0.0030 | 1.22 | 0.2475  |
|       |           |          |        |      | 0.2492  |
|       |           |          |        |      | 0.2434  |
| Std 5 | 1.541     | 0.3748   | 0.0006 | 0.17 | 0.3741  |
|       |           |          |        |      | 0.3750  |
|       |           |          |        |      | 0.3753  |
| Std 6 | 6.162     | 1.4506   | 0.0011 | 0.07 | 1.4503  |
|       |           |          |        |      | 1.4496  |
|       |           |          |        |      | 1.4517  |

Two samples of tap water from different sources, A and B, were prepared as described in the reference 1. The absorbance was measured and the concentration of nitrate determined from the calibration curve. The results are shown in Table 2.

**Table 2.** Raw data and statistics of Water samples

| Std | Conc mg/L | Mean Abs | SD     | %RSD | Raw Abs |
|-----|-----------|----------|--------|------|---------|
| A   | 0.145     | 0.0510   | 0.0009 | 1.78 | 0.0520  |
|     |           |          |        |      | 0.0504  |
|     |           |          |        |      | 0.0506  |
| B   | 0.709     | 0.1825   | 0.0025 | 1.36 | 0.1797  |
|     |           |          |        |      | 0.1838  |
|     |           |          |        |      | 0.1841  |

## Discussion

The 3 replicates for each standard and sample, shown in Tables 1 and 2, are reproducible within instrumental uncertainty, which demonstrates the high precision possible using fiber optics on the Cary 50/60. There is negligible solution carry over between samples, washing with only distilled water for approximately 5 seconds.

The time taken to measure 24 solutions of 6 standards and 2 samples, each with 3 replicates, was approximately 5 minutes. This time included washing the probe with de-ionized water in between readings and drying with a tissue. Measurements with the dip probe are significantly faster and easier than using a conventional cuvette.

## Conclusion

The quartz fiber optic dip probe on the Cary 50/60 is highly precise and efficient for measuring the nitrate content in water. The time taken to measure 24 solutions is faster than using a cuvette, which makes the technique an attractive alternative for routine analytical measurements.

## Reference

1. D. Eaton, L. S. Clesceri and A. E. Greenberg, *Standard Methods for the Examination of Water and Wastewater*, 19<sup>th</sup> Edition, American Public Health Association, Washington, 1995, p4-85.
2. P. A. Liberatore, *UV-Instrument At Work; Automated nitrate analysis of water*, No. 59, Agilent Australia Pty. Ltd, Australia, 1993.

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